

Onsager's Variational Principle in Soft Matter Physics

Masao Doi
Department of Applied Physics
University of Tokyo

Aim of my talk

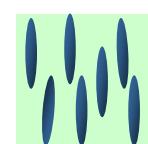
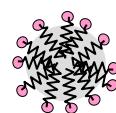
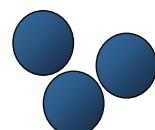
Soft Matter

Polymer

Colloids

Surfactants

Liquid crystals



Consisting of very large unit

- Strongly perturbed from equilibrium state by weak force
- Relaxation to equilibrium is very slow



Strongly non-linear and non-equilibrium

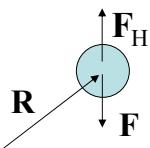
The PDEs which describe the non-linear dynamics of soft matter can be conveniently derived by the Onsager's variational principle.

Outline

1. Variational principle in Stokesian hydrodynamics
2. Onsager's variational principle in irreversible thermodynamics
3. Applications
 1. Diffusion in colloidal solutions
 2. Liquid crystals (Leslie-Ericksen theory)
4. Summary

Variational Principle in
Stokesian Hydrodynamics

Sedimentation of a Particle in a Viscous Fluid



$$\mathbf{F} = \mathbf{mg} = -\frac{\partial \mathbf{U}}{\partial \mathbf{R}} \quad \text{potential force}$$

$$\mathbf{F}_h = -\zeta \dot{\mathbf{R}} \quad \text{frictional force}$$

In a steady state minimize

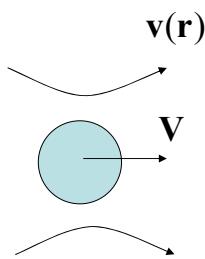
$$\zeta \dot{\mathbf{R}} = -\frac{\partial U}{\partial \mathbf{R}} \quad \Leftrightarrow \quad \mathbf{R} = \frac{1}{2} \zeta \dot{\mathbf{R}}^2 + \frac{\partial U}{\partial \mathbf{R}} \bullet \dot{\mathbf{R}}$$

↑
↑

dissipation function
 \dot{U}

The principle of the least dissipation of energy

Friction Constant for a Sphere



$$\mathbf{F}_H = -\zeta \dot{\mathbf{R}}$$

$$\nabla^2 \mathbf{v} = \nabla p$$

$$\nabla \bullet \mathbf{v} = 0$$

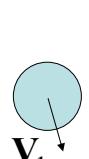
$\mathbf{v} \equiv \mathbf{V}$ at the surface

$$\mathbf{v} = 0 \quad \text{at infinity}$$

$$\mathbf{F}_H = \int dS \boldsymbol{\sigma} \bullet \mathbf{n} = -\zeta \mathbf{V}$$

$$\zeta = 6\pi\eta a$$

Sedimentation of Two Interacting Particles in a Viscous Fluid



$$U = mg \bullet R_1 + mg \bullet R_2 + U_{int}(R_1, R_2)$$

$$\zeta_{11}\dot{R}_1 + \zeta_{12}\dot{R}_2 = -\frac{\partial U}{\partial R_1}$$

$$\zeta_{21}\dot{R}_1 + \zeta_{22}\dot{R}_2 = -\frac{\partial U}{\partial R_2}$$

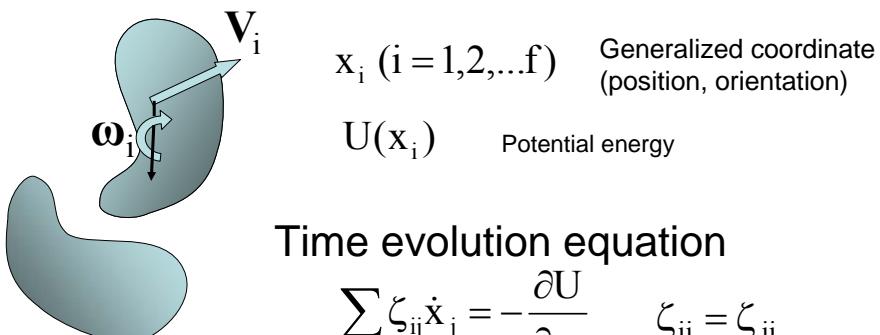
$$\zeta_{ij} = \zeta_{ji}(R_1, R_2)$$

$\zeta_{ij} = (\zeta_{ji})^t$ Helmholtz's reciprocal relation

$$R = \frac{1}{2} \sum \zeta_{ij} : \dot{R}_i \dot{R}_j + \sum \frac{\partial U}{\partial R_i} \bullet \dot{R}_i$$

Variational Principle for Particle Dynamics in Viscous Fluids

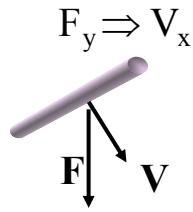
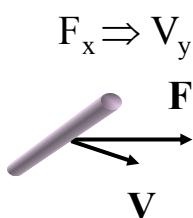
Interacting particle moving in a viscous fluid



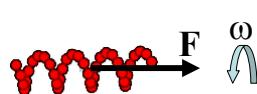
$$R = \frac{1}{2} \sum \zeta_{ij} \dot{x}_i \dot{x}_j + \sum \frac{\partial U}{\partial x_i} \dot{x}_i$$

Lorentz reciprocal theorem

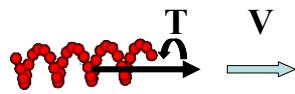
Reciprocal Relation is Not a Trivial Relation



$$V_y / F_x = V_x / F_y$$



$$\omega = \mu_{tr} F$$



$$V = \mu_{tr} T$$

Proof of the reciprocal relation

1. Proof by hydrodynamics (Lorentz)
2. Proof by phenomenology (Onsager)
3. Proof by statistical mechanics (Kubo et al)

Hydrodynamic proof

$$\begin{array}{ccccc} \text{Configurational change} & & \text{Fluid velocity} & & \text{Frictional force} \\ \dot{x}_i^{(1)} & \longrightarrow & v^{(1)} & \longrightarrow & F_i^{(1)} \\ \dot{x}_i^{(2)} & \longrightarrow & v^{(2)} & \longrightarrow & F_i^{(2)} \end{array}$$

We prove

$$\sum \zeta_{ij} \dot{x}_i^{(1)} \dot{x}_j^{(2)} = \sum F_i^{(1)} \dot{x}_i^{(2)} = \sum \zeta_{ij} \dot{x}_i^{(2)} \dot{x}_j^{(1)}$$

$$\begin{aligned} \sum F_i^{(1)} \dot{x}_i^{(2)} &= - \int dS G_{i\alpha}(\mathbf{r}) \sigma_{\alpha\beta}^{(1)} n_\beta \dot{x}_i^{(2)} \\ &= - \int dS \sigma_{\alpha\beta}^{(1)} n_\beta v_\alpha^{(2)} = - \int dS \sigma_{\alpha\beta}^{(2)} n_\beta v_\alpha^{(1)} = \sum F_i^{(2)} \dot{x}_i^{(1)} \\ \sigma_{\alpha\beta}^{(1)} &= \eta [\partial_\beta v_\alpha^{(1)} + \partial_\alpha v_\beta^{(1)}] - p^{(1)} \delta_{\alpha\beta} \end{aligned}$$

$$\begin{aligned} - \int dS \sigma_{\alpha\beta}^{(1)} n_\beta v_\alpha^{(2)} &= \int d\mathbf{r} \partial_\beta (\sigma_{\alpha\beta}^{(1)} v_\alpha^{(2)}) \\ &= \int d\mathbf{r} [v_\alpha^{(2)} \partial_\beta \sigma_{\alpha\beta}^{(1)} + \sigma_{\alpha\beta}^{(1)} \partial_\beta v_\alpha^{(2)}] \\ &= \int d\mathbf{r} \sigma_{\alpha\beta}^{(1)} \partial_\beta v_\alpha^{(2)} = (1/2) \int d\mathbf{r} [\partial_\beta v_\alpha^{(1)} + \partial_\alpha v_\beta^{(1)}] [\partial_\beta v_\alpha^{(2)} + \partial_\alpha v_\beta^{(2)}] \end{aligned}$$

Onsager's proof 1/2

Assume that the friction matrix depends only on \mathbf{x} and independent of \mathbf{U}

$$-\sum \zeta_{ij} \dot{x}_j - \frac{\partial U}{\partial x_i} + F_{ri}(t) = 0$$

Assume that $x_i(t)$ is fluctuating around x_{i0}

$$\xi_i(t) = x_i(t) - x_{i0}$$

$$U(\xi) = \frac{1}{2} \sum k_{ij} \xi_i \xi_j$$

$$-\sum \zeta_{ij} \dot{\xi}_j - \sum k_{ij} \xi_j + F_{ri}(t) = 0$$

We can apply a hypothetical force to balance $-\partial U / \partial x_i$

$$\text{Calculate } \langle \xi_i(t + \Delta t) \xi_j(t) \rangle$$

$$\text{Use the symmetry } \langle \xi_i(t) \xi_j(t') \rangle = \langle \xi_j(t) \xi_i(t') \rangle$$

$$\downarrow$$

$$\zeta_{ij} = \zeta_{ji}$$

Time correlation at equilibrium has time reversal symmetry:
 $\langle A(t)B(0) \rangle = \langle B(t)A(0) \rangle$

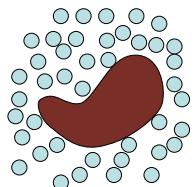
Onsager's proof 2/2

$$-\sum \zeta_{ij} \dot{\xi}_j - \sum k_{ij} \xi_j + F_{ri}(t) = 0$$

$$\dot{\xi}_i = -\sum \alpha_{ij} \xi_j + v_{ri}(t) \quad \alpha_{ij} = \sum (\zeta^{-1})_{im} k_{mj}$$

$$\begin{aligned} \xi_i(t + \Delta t) &= \xi_i(t) - \Delta t \sum \alpha_{ij} \xi_j(t) + \int_t^{t+\Delta t} dt' v_{ri}(t') \\ \langle \xi_i(t + \Delta t) \xi_j(t) \rangle &= \langle \xi_i(t) \xi_j(t) \rangle - \Delta t \sum \alpha_{ik} \langle \xi_k(t) \xi_j(t) \rangle \quad \langle \xi_i(t) \xi_j(t) \rangle = (k^{-1})_{ij} k_B T \\ &= k_B T \left[(k^{-1})_{ij} - \Delta t \sum \alpha_{ik} (k^{-1})_{kj} \right] \\ &= k_B T \left[(k^{-1})_{ij} - \Delta t \sum (\zeta^{-1})_{ij} \right] \\ \langle \xi_i(t + \Delta t) \xi_j(t) \rangle &= \langle \xi_j(t + \Delta t) \xi_i(t) \rangle \\ &\downarrow \\ \zeta_{ij} &= \zeta_{ji} \end{aligned}$$

Proof by statistical mechanics 2/2

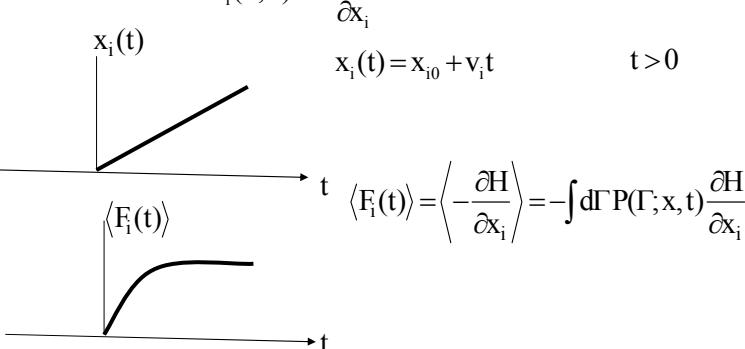


$H(\Gamma; x)$

- Parameters representing the configuration of Brownian particles
- Phase space variables representing the configuration of solvent molecules

Force exerted on the particle by fluid molecules

Assume



Proof by statistical mechanics

$$\langle F_i(t) \rangle = \int d\Gamma \left(-\frac{\partial H}{\partial x_i} \right) P(\Gamma; x, t)$$

$t < 0 \quad P(\Gamma; x, t) \propto \exp[-\beta H(\Gamma; x_0)]$

$$\langle F_i \rangle = -\frac{\partial A(x_0)}{\partial x_{i0}} = \bar{F}_i(x_0) \quad A(x_0) = -\frac{1}{\beta} \ln \int d\Gamma e^{-\beta H(\Gamma; x_0)}$$

$t > 0$ Perturbative solution for $\frac{\partial P}{\partial t} = (L + L')P$ gives the following result.

$$\langle F_i(t) \rangle = -\frac{\partial A}{\partial x_i} - \int_0^t dt' \sum \zeta_{ij}(t') v_j \quad \dot{\zeta}_{ij}(t) = \frac{1}{k_B T} \langle F_i(t) F_j(0) \rangle_0$$

Especially for $t > \tau_F$

$$\langle F_i(t) \rangle = -\frac{\partial A}{\partial x_i} - \sum \zeta_{ij} v_j \quad \zeta_{ij} = \frac{1}{k_B T} \int_0^\infty dt' \langle F_i(t') F_j(0) \rangle_0$$

Onsager's Variational Principle in Irreversible Thermodynamics

Onsager's Reciprocal Relation

Onsager 1931

$(x_1, x_2, x_3 \dots)$ State variables describing non-equilibrium state

$$S = S(x_1, x_2, x_3 \dots)$$

$$\dot{x}_i = \sum L_{ij} \frac{\partial S}{\partial x_j}$$

Onsager's reciprocal relation

$$L_{ij} = L_{ji}$$

Can be proven by time reversal symmetry in the fluctuation at equilibrium state

Onsager's Variational Principle

$$x_i = \sum L_{ij} \frac{\partial S}{\partial x_j}$$

Onsager 1931

$$L_{ij} = L_{ji}$$

$$O = -\frac{1}{2} \sum (L^{-1})_{ij} \dot{x}_i \dot{x}_j + \sum \frac{\partial S}{\partial x_i} \dot{x}_i \quad \text{variational principle}$$

If T (temperature) is constant:

$$R = \frac{1}{2} \sum \zeta_{ij} \dot{x}_i \dot{x}_j + \sum \frac{\partial A}{\partial x_i} \dot{x}_i$$

$A = A(x_i)$
free energy of the system

Application 1

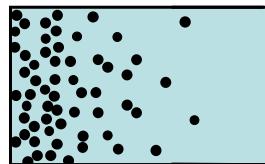
Diffusion in Concentrated Colloidal Particles

Recipe to Make Kinetic Equations

- (1) Choose proper state variables $(x_1, x_2, x_3 \dots)$
- (2) Construct the free energy function $A = A(x_i)$
- (3) Construct the dissipation function $(1/2) \sum \zeta_{ij} V_i V_j$
- (4) Minimize $R = \frac{1}{2} \sum \zeta_{ij} \dot{x}_i \dot{x}_j + \sum \frac{\partial A}{\partial x_i} \dot{x}_i$
$$\sum \zeta_{ij} \dot{x}_j = - \frac{\partial A}{\partial x_i}$$

Diffusion of Colloidal Particles

Diffusion of particles in quiescent solution



State variable $n(x, t)$ Volume fraction of particles

Our objective is to determine $\dot{n}(x, t)$

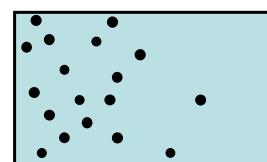
It is more convenient to consider $v_p(x, t)$ Solute velocity

$$\dot{n} = -\frac{\partial}{\partial x}(v_p n)$$

Case of Dilute Solution

Free Energy of the system

$$A[n(x)] = \int dx [k_B T n(x) \ln n(x)]$$



$$\begin{aligned}\dot{A} &= \int dx \dot{n} (k_B T \ln n + k_B T) \\ &= - \int dx \frac{\partial v_p n}{\partial x} (k_B T \ln n + k_B T) \\ &= \int dx v_p n \frac{\partial}{\partial x} (k_B T \ln n + k_B T) \\ &= k_B T \int dx v_p \frac{\partial n}{\partial x}\end{aligned}$$

Variational Calculation

$$R = \frac{1}{2} \int dx \zeta n v_p^2 + k_B T \int dx v_p \frac{\partial n}{\partial x} \quad \zeta = 6\pi\eta a$$

$$\left. \begin{array}{l} \zeta n v_p + k_B T \frac{\partial n}{\partial x} = 0 \\ \dot{n} = -\frac{\partial}{\partial x} (v_p n) \end{array} \right\} \quad \begin{aligned} \frac{\partial n}{\partial t} &= D \frac{\partial^2 n}{\partial x^2} \\ D &= \frac{k_B T}{\zeta} \end{aligned}$$

Case of Concentrated Solution

$$\phi = n \frac{4\pi}{3} a^3$$

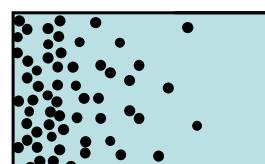
$$A[\phi] = \int dr f(\phi)$$

$$\dot{A} = \int dr f'(\phi) \dot{\phi}$$

$$= - \int dr f'(\phi) \nabla \bullet (\phi v_p)$$

$$= \int dr v_p \bullet \phi \nabla f'(\phi)$$

$$= \int dr v_p \bullet \underbrace{\nabla \Pi}_{\leftarrow \quad \text{Osmotic pressure}} \quad \Pi(\phi) = \phi f' - f$$



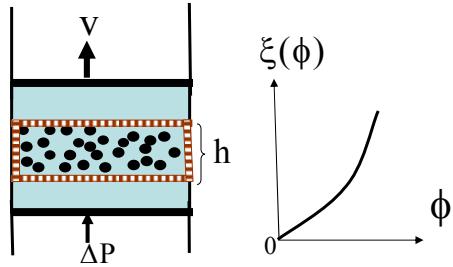
$$\nabla \Pi = (\nabla \phi) f' + \phi \nabla f' - \nabla f = \phi \nabla f'$$

Dissipation Function

$$W = \int d\mathbf{r} \xi(\phi) v_p^2$$

↑
Friction coefficient / volume

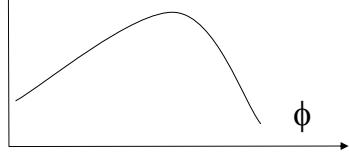
$$\Delta P / h = \xi v$$



Variational Calculation

$$R = \frac{1}{2} \int d\mathbf{r} \xi(\phi) v_p^2 + \int d\mathbf{r} \mathbf{v}_p \bullet \nabla \Pi$$

$$\left. \begin{aligned} \mathbf{v}_p &= -\frac{\nabla \Pi}{\xi(\phi)} \\ \dot{\phi} &= -\nabla \bullet (\phi \mathbf{v}_p) \end{aligned} \right\} \quad \begin{aligned} \frac{\partial \phi}{\partial t} &= \nabla (D \nabla \phi) \\ D(\phi) &= \frac{\phi}{\xi(\phi)} \frac{\partial \Pi}{\partial \phi} \end{aligned}$$



Reciprocal Relation in Diffusion Equation

Onsager's kinetic equation

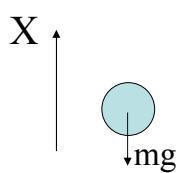
$$\dot{X}_i = \sum L_{ij} \frac{\partial A}{\partial X_j} \quad L_{ij} = L_{ji}$$

Diffusion equation

$$\frac{\partial \phi}{\partial t} = \nabla(D \nabla \phi)$$

$$\frac{\partial \phi}{\partial t} = - \int d\mathbf{r} \int d\mathbf{r}' \mu(\mathbf{r}, \mathbf{r}') \frac{\delta A}{\delta \phi(\mathbf{r}')} \quad \mu(\mathbf{r}, \mathbf{r}') = \mu(\mathbf{r}', \mathbf{r})$$

Forces needed to change external parameters



$$R = \frac{1}{2} \zeta \dot{X}^2 + mg \dot{X}$$

$$F = \frac{\partial R}{\partial \dot{X}} = \zeta \dot{X} + mg$$

In general

$$R = R(\dot{x}, x, p, \dot{p})$$

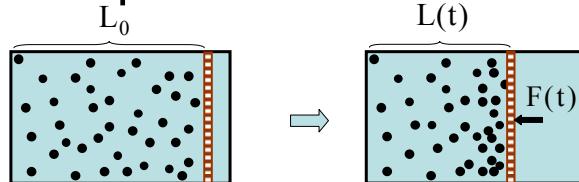
Internal variables

External parameters

The forces needed to change the external parameter is

$$F = \frac{\partial R}{\partial \dot{p}}$$

Forces needed to move the semipermeable membrane



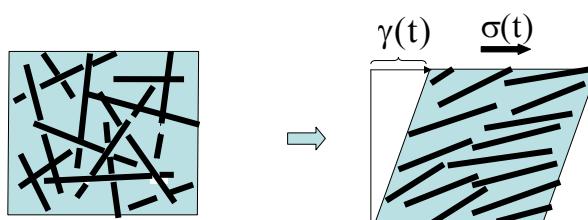
$$A = \int_0^L dx k_B T n(x) \ln n(x)$$

$$\dot{A} = - \int_0^L dx k_B T v_p \frac{\partial n}{\partial x} + \dot{L} n(L) k_B T$$

$$W = \int_0^L dx \zeta n v_p^2 + \zeta_m \dot{L}^2$$

$$F(t) = \frac{\partial R}{\partial L} = \zeta_m \dot{L} - k_B T n(L)$$

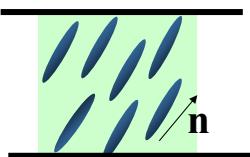
Rheology of rodlike polymers



$$\sigma(t) = \frac{\partial R}{\partial \dot{\gamma}}$$

Application 2 Liquid Crystals

Flow of Liquid Crystals



$\mathbf{n}(\mathbf{r}, t)$ director

$\mathbf{v}(\mathbf{r}, t)$ velocity

$$\tilde{\dot{\mathbf{n}}}_\alpha = \dot{\mathbf{n}}_\alpha - \omega_{\alpha\beta} \mathbf{n}_\beta$$

$$\omega_{\alpha\beta} = \frac{1}{2} (\partial_\beta v_\alpha - \partial_\alpha v_\beta)$$

$$\dot{\varepsilon}_{\alpha\beta} = \frac{1}{2} (\partial_\beta v_\alpha + \partial_\alpha v_\beta)$$

Nemato-Hydrodynamics

$$W = \frac{A_1}{2} (n_\alpha n_\beta \dot{\epsilon}_{\alpha\beta})^2 + \frac{A_2}{2} \dot{\epsilon}_{\alpha\beta} \dot{\epsilon}_{\alpha\beta} + \frac{A_3}{2} n_\mu n_\nu \dot{\epsilon}_{\alpha\mu} \dot{\epsilon}_{\beta\nu}$$
$$+ \frac{A_4}{2} (\tilde{n}_\alpha)^2 + \frac{A_5}{2} \tilde{n}_\alpha \dot{\epsilon}_{\alpha\beta} n_\beta$$
$$A = \frac{1}{2} K_1 (\nabla \bullet \mathbf{n})^2 + \frac{1}{2} K_2 (\mathbf{n} \bullet \nabla \times \mathbf{n})^2 + \frac{1}{2} K_3 (\mathbf{n} \times \nabla \times \mathbf{n})^2$$
$$\dot{U} = \frac{\delta F}{\delta n_\alpha} \dot{n}_\alpha - \frac{\partial F}{\partial n_{\alpha,\beta}} (\partial_\mu n_\alpha) (\partial_\beta v_\mu)$$

➡ Leslie-Ericksen equation

Conclusion

- Many kinetic equations in soft matter physics can be derived from the Rayleigh-Onsager's variational principle
 - Brownian motion of colloidal particles
 - Phase separation kinetics in solutions
 - Swelling kinetics in gels
 - Flow in liquid crystals
 - Diffusio-phoresis, electro-phoresis etc
- The variational principle is simple and easy to use.

The Onsager's variational principle is a general base of soft matter physics.